

PROBLEM I

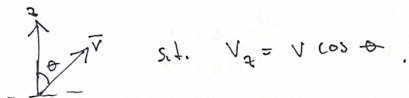
"Find the number of impacts of gas molecules on unit area of the wall per unit time"

Suppose we have some container with molecules hitting the walls. Take a snapshot of the wall:



with side length 1. Then let the z-axis be \odot .

Clearly, only $v_z > 0$ hit the wall. Let the velocity \vec{v} make an angle θ with z-axis:



The nr of collisions of molecules per unit time & area of the wall surface is given by:

$$dN_v = \frac{N}{V} \left(\frac{m}{2\pi T} \right)^{3/2} \exp \left[-m(v_x^2 + v_y^2 + v_z^2) / 2T \right] \times v_z dv_x dv_y dv_z$$

This can be written in the velocity space where

$$dv_x dv_y dv_z = v^2 \sin\theta dv d\theta d\phi. \text{ Hence, we get}$$

$$dN_v = \frac{N}{V} \left(\frac{m}{2\pi T} \right)^{3/2} e^{-m^2/2T} v^3 \sin\theta \cos\theta dv d\theta d\phi. \text{ As requested,}$$

we are interested in the nr of impacts between θ and $\theta + d\theta$, translated; integrate all speeds v from 0 to ∞ and ϕ from 0 to 2π .

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$$dV_{\vec{v}} = \frac{N}{V} \left(\frac{m}{2\pi T} \right)^{3/2} \sin\theta \cos\theta d\theta \int_0^{2\pi} d\phi \int_0^{\infty} e^{-mv^2/2T} v^3 dv$$

$$\text{let } I = \int_0^{\infty} e^{-mv^2/2T} v^3 dv = \left[\begin{array}{l} u = \frac{m}{2T} v^2 \\ du = \frac{m}{T} v dv \end{array} \right]$$

$$= \frac{2T^2}{m^2} \int_0^{\infty} u e^{-u} du = \frac{2T^2}{m^2}$$

$$\text{So } dV_{\vec{v}} = \frac{N}{V} \left(\frac{m}{2\pi T} \right)^{3/2} \sin\theta \cos\theta d\theta \cdot 2\pi \cdot \frac{2T^2}{m^2}$$

$$= \frac{N}{V} \left(\frac{2T}{m\pi} \right)^{3/2} \sin\theta \cos\theta d\theta$$

What we notice here is that surfaces only see a version of the velocity distribution despite the fact that velocities in the gas are isotropic.